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# A review on battery thermal management and its digital improvement based Cyber Hierarchy and Interactional Network (CHAIN)

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#### 20 Abstract

The temperature has a significant influence on the performance of Lithium-ion 21 22 batteries (LiBs). Meanwhile, the heat generated accumulation in the battery can trigger the battery's thermal runaway. Hence, the battery thermal management system (BTMS) 23 is essential to ensure the safe and reliable operation of the battery. This paper 24 comprehensively reviewed key technologies of BTMS and proposed a novel digital 25 solution to improve the battery system performance. Firstly, the heat generation 26 mechanisms and the thermal models were reviewed. Then this paper focuses on a 27 28 review of liquid cooling thermal management methods. simultaneously, the solidliquid phase change gas-liquid phase change is also summarized in the phase change 29 30 material cooling system, and a summary is presented for the existing novel phase 31 change cooling systems. Additionally, the preheating methods to heat the LiBs at low 32 temperatures and the emergency battery thermal barriers upon thermal runaway were discussed. Eventually, a new approach for the BTMS leveraging from the Cyber 33 Hierarchy and Interactional Network (CHAIN) framework is indicated and constructed 34 the digital twin reflecting physical battery to improve the LiBs temperature control 35 strategies. Besides, this paper provides a new direction for the design of BTMS through 36 vehicle-side sensing, edge computing, and cloud-based digital twin three levels. 37

#### 1 key words

2 Lithium-ion battery; Heat generation mechanism; Battery thermal model; Battery

3 thermal management system; Emergency battery thermal barrier; Cyber Hierarchy and

4 Interactional Network; Digital-twin

5

Nomencl	ature		
q	Heat generation rate, j·s <sup>-1</sup>	ρ	Density, kg·m <sup>-3</sup>
Ι	Electric current, A	κ <sub>D</sub>	Electrolyte phase diffusion conductivity, $\Omega^{-1}$ cm <sup>-1</sup>
U	Open-circuit voltage, V	λ	Thermal conductivity, W·m <sup>-1</sup> ·K <sup>-1</sup>
V	Cell voltage, V	α	Degree of conversion
Т	Temperature, K	Superscripts	,
j <sup>li</sup>	Butler-Volmert current density, A cm <sup>-3</sup>	eff	Effective
<i>as</i>	Active surface area per electrode unit volume, cm <sup>2</sup> cm <sup>-3</sup>	Subscripts	
R	Universal gas constant, J mol <sup>-</sup> <sup>1</sup> K <sup>-1</sup>	е	Electrolyte phase
С	Concentration of lithium ions, mol cm <sup>-3</sup> , also dimensionless concentration	S	Solid phase
R <sub>f</sub>	Film resistance on the electrodes surface, $\Omega cm^2$	п	negative
A	Electrode plate Area, cm <sup>2</sup>	р	positive
V	Cell potential, V	sei	solid electrolyte interface
x	the coordinate across the	ne	negative-electrolyte
	thickness of the electrode, cm		
r	Radial coordinate, cm	Abbreviation	15
t	Time. s	AC	Alternating current
D	Diffusion coefficient of lithium species, cm <sup>2</sup> s <sup>-1</sup>	BTMS	Battery thermal management system
R <sub>s</sub>	Solid active material particle radius, cm	CHAIN	Cyber Hierarchy and Interactional Network
F	Faraday's Constant, C mol-1	DC	Direct current
$t_0^+$	Transference number of lithium	EV	Electric vehicle
	ion		
σ	Conductivity of solid active material, $\Omega^{-1}$ cm <sup>-1</sup>	EIS	Electrochemical impedance spectroscopy
L	The thickness of cell, cm	EBTB	Emergency battery thermal barrier
$C_p$	Heat capacity, J·kg <sup>-1</sup> ·K <sup>-1</sup>	HEV	Hybrid electric vehicles
v	Velocity, m·s <sup>-1</sup>	HP	Heat pipe

Н	Total enthalpy, kJ. k g <sup>-1</sup>	HDMR	High dimensional model representation
W	specific content in jellyroll,	IOT	Internet of things
	kg/m <sup>3</sup>		
Q	Heat generation rate, $j \cdot s^{-1}$	LiB	Lithium-ion battery
а	frequency factor, 1/s	OHP	Oscillating heat pipe
Ζ	Dimensionless solid electrolyte	OCV	Open-circuit voltage
	interface (SEI) thickness		
Ε	activation energy, J/mol	OBC	Olefin block copolymer
b	battery	P2D	Pseudo 2 dimensional
Greek symbols		PCM	Phase change material
$\alpha_a, \alpha_c$	Anodic and catholic charge	PDE	Partial differential equations
	transfer coefficients		
η	Surface overpotential of an	SOC	State of charge
	electrode reaction, V		
$\phi$	Volume averaged electrical	SEI	Solid electrolyte interface
	potential, V		
$\varepsilon_e$	Electrolyte phase volume	TR	Thermal runaway
	fraction		
σ	Conductivity of solid active	TEG	Thermoelectric generators
	material, $\Omega^{-1}$ cm <sup>-1</sup>		
δ	Thickness, cm	TEC	Thermoelectric cooler
κ	Electrolyte phase ionic		
	conductivity, $\Omega^{-1}$ cm <sup>-1</sup>		

1

#### 2 **1. Introduction**

With fossil fuel leading to environmental pollution and the exhaustion of non-3 4 renewable energy sources, increasing interest has focused on electric vehicles (EVs) 5 and hybrid electric vehicles (HEVs)[1]. Lithium-ion batteries are the most popular power sources in EVs due to their advantages, such as high energy density, high 6 power density, and low self-discharge rate[2-4]. Nevertheless, the performance and 7 8 safety of LiBs are very sensitive to temperature [5]. Low temperature significantly 9 reduces the performance of LIBs. Therefore, there are still some problems to apply 10 LiBs in cold climate areas, such as reduced driving range and battery life [6–8]. High temperature also has some detrimental effects on battery performances including 11 12 accelerated capacity, power, and energy losses [9–11]. In addition, the more serious 13 consequence of high temperature is thermal runaway (TR) which could damage the

vehicle and endanger passengers' life. So the BTMS is of great importance in
 achieving high performance and high safety of LiBs.

3 The battery thermal management has two critical issues, heat generation and heat dissipation. In terms of heat generation, different models have been developed, such 4 as electro-thermal model, electrochemical-thermal model and lumped model. For heat 5 dissipation, different BTMS are available, such as the air based, liquid based, phase 6 7 change material (PCM) based cooling and hybrid cooling system. In general, the heat transfer medium used in BTMS mainly include air, liquid, PCM, and thermoelectric 8 9 element. Air based cooling is the simplest method, which uses passive or active air flow to keep the battery within the optimal temperature range. There are numerous 10 11 optimal design strategies to improve the air cooling efficacy. Optimizing the cell spacing can achieve more uniform airflow rates for the parallel air cooling BTMS[12], 12 which obviously improves the cooling performance. The inlet and outlet locations for 13 air based cooling system can be optimized to enhance the cooling performance. For 14 example, experimental data indicate that the symmetrical inlet and outlet set in the 15 16 middle of the plenums can achieve higher cooling efficacy for symmetrical BTMS[13]. Though air cooling systems have been widely used, the thermal 17 conductivity of air is lower than other cooling medium, which limits its application in 18 automotive. Fluid medium has higher thermal conductivity than air. Therein, the 19 commonly used liquid mediums are water and ethylene glycol. Shang adopted the 20 water and ethylene glycol as the coolant to design the BTMS for battery pack. 21 Wherein, the flow rate is controlled to adjust the cooling rate. The BTMS has 22 effectively restricted the maximum temperature rise of the battery[14]. In describing 23 24 batteries, the discharge current is often expressed as a C-rate to normalize against battery capacity, which is usually very different between batteries. C-rate is used to 25 26 measure the rate at which a battery is discharged relative to its maximum capacity. Chaithanya et al. conducted the cooling research on a 20Ah LiFePO4 battery using 27 28 COMSOL software. The experiments were carried out by placing the cell between two mini-channel cold plates and conducting bi-directional coupled electrochemical-29

thermal simulations at different experimental conditions of discharge multiplicity (1-1 4C) and coolant inlet temperature (15-35°C). Meanwhile, temperatures were collected 2 at ten different locations on the cell surface, and it was found that the measured and 3 predicted values were very close to each other and that the temperature difference 4 between the two was due to the assumption of uniform heat generation in the Li-ion 5 6 battery model (P2D). The experimental results show that the Li-ion battery model can be used to design battery management systems[15]. Sandeep Dattu Chitta et al. 7 8 comparatively investigated temperature prediction accuracy for a 20Ah LiFePO4 battery sandwiched between mini-channel cooling plates based on two different 9 10 simplified cell models (Lumped and 1D electrochemical) models. Ten other locations on the cell surface were selected under different experimental conditions of discharge 11 12 rate (1-4C) and coolant inlet temperature  $(15-35^{\circ}C)$ , and the temperature changes during discharge were compared. The experimental results found that the difference 13 between the Lumped model predicted temperature and the 1D electrochemical 14 15 predicted temperature was 2.1% at the 1C discharge rate; however, the difference 16 increased to 16.4% at the 4C discharge rate, which is because the Lumped model 17 ignores the cell's internal chemistry and is therefore recommended for the initial design of the BTMS[16]. There are some ways to improve the fluid thermal 18 19 conductivity, e.g., by adding solid particles to the coolant. Experiments indicated that 20 adding the different volumetric fractions of nanoparticle (Fe<sub>2</sub>O<sub>3</sub>) to the coolant (water and ethylene glycol) can improve fluids thermal conductivity[17]. The advantages of 21 liquid BTMS include high cooling efficiency, high reliability and high heat capacity. 22 However, one major disadvantage is that it increases the cost and complexity of 23 24 BTMS due to the pump and other accessories. This can be problematic for small electric cars with limited interior space. 25

PCM-based BTMS can be used to address the liquid cooling issues of cost and complexity, because it doesn't require an extra power source. PCM is generally filled in the gaps between cells[18,19]. V.G. Choudhari et al. investigated the thermal performance of a battery module containing 5x5 Li-ion cells arranged in series and

parallel using phase change materials. Four different fin structure layouts, such as 1 Type I, Type II, Type III, and Type IV, were experimentally proposed to improve the 2 cells' internal heat dissipation, and the maximum and average temperature 3 distributions in the battery pack were analyzed. The results show that the type III fin 4 structure can minimize the internal heat build-up. At the same time, the experiments 5 6 found that the convection effect can reduce the maximum temperature inside the cell and restore the melting fraction of PCM by investigating the impact of convection on 7 8 the thermal performance of the cell[20]. When the battery temperature is high, PCM can absorb heat from the cells to limit the maximum temperature rise. The stored heat 9 10 in the PCM can be released when the battery's temperature drops below liquid-solid phase transition point, e.g., as the vehicle travels at night. The advantage of the PCM-11 based BTMS include broad range of suitable phase change temperature, no consume 12 parasitic energy and low cost[21]. However, one disadvantage is that the latent heat 13 of PCM is limited, so it cannot absorb too much heat. Battery spacing is a key to the 14 performance of PCM based BTMS. As the battery spacing increases, more PCM can 15 16 be filled in the gaps, which can improve the thermal control efficiency. The hybrid 17 BTMS design that combines PCM/oscillating heat pipe (OHP) also can solve the limited latent heat of PCM[22]. Another disadvantage of PCM-based BTMS is that 18 the leakage of melted PCM also brings some detriment, such as environmental 19 20 pollution and battery TR[23].

Thermoelectric element based BTMS has gradually attracted people's attention 21 owing to the high efficiency. No chemical coolant is used, which avoids leakage 22 problems. A calibrated model of thermoelectric element based BTMS was developed 23 in [24] to investigate its performance. The result indicated the thermoelectric element 24 cooling method can cool the cell with high efficiency and keep a more homogeneous 25 26 temperature distribution inside the battery module. Cai et al. adopted thermoelectric element cooling in combination with air cooling or liquid cooling that could achieve 27 effectively and actively cooled[25]. 28

In general, BTMS has a three-layer design concept. Firstly, it should ensure the 1 2 battery operates within the optimal temperature range. Secondly, it should be able to sense the critical edge of battery malfunctions and send the alert information. Finally, 3 once the thermal runaway happens, the relevant measures can suppress thermal 4 runaway propagation effectively[26]. Kanishka Vikram Purohit et al. investigated the 5 6 soft sensor for a battery pack system based on a two-layer feed-forward artificial neural network, by which the soft sensor was able to accurately estimate the state of 7 8 charge (SOC), state of energy (SOE), and power loss of the battery pack. The prediction accuracy of the soft sensor is also compared with linear or non-linear 9 10 regression models and parametric structure models for system identification. The accuracy of the soft sensor is 99.96%, 99.96%, and 99.99% for the battery pack SOC, 11 SOE, and PL test datasets, respectively. It shows that the soft sensor has a higher 12 prediction accuracy[27]. However, the most previous research on battery thermal 13 14 management methods has focused on model simulation and experimental validation to optimize the thermal management system materials, structure and physical 15 16 parameters (flow rate, latent heat, viscosity etc.). They have not further combined with the existing artificial intelligence, big data, models, and control algorithms to 17 realize a whole life cycle battery thermal management strategy. Therefore, on the 18 19 basis of summarizing the existing thermal management methods, this paper proposes 20 a new digital solution, namely the full lifecycle BTMS based on CHAIN, which can be achieved by machine learning, big data, and model fusion etc. This review 21 summarizes the heat generation mechanisms, thermal models, and thermal 22 23 management methods of LiBs, and proposes a novel strategy that combines battery 24 thermal management with cloud computing.

This paper is organized as follow. The heat generation mechanism and heat transfer model of LiBs are presented in Section 2. The three major applications of battery thermal management in EVs are reviewed in Section 3. The major applications are classified as cooling methods, battery preheating at low temperature, and emergency battery thermal barrier, which will be given in Section3.1, 3.2, and 3.3 separately. Section 4 presents a novel approach for the full-lifetime BTMS leveraging
 from the CHAIN framework, which helps to perfect the temperature control strategy
 of LiBs. Finally, the conclusions and directions for future research are given in
 Section 5.

#### 5 2. Heat generation mechanisms and thermal models of LiBs

Heat generated inside LiBs mainly includes reaction heat, polarization heat, and
joule heat[28]. Under abuse conditions, e.g., when the battery's temperature is too high,
a large amount of heat can also be generated by side reactions, such as electrode
decomposition[29]. In the worst case, with the accumulation of heat generated from
the side reactions, the battery temperature will rise sharply, which eventually will result
in TR.





13

FIGURE 1 Heat generation mechanisms and thermal models of LiBs[30–34]

A battery thermal model describes the heat generation and dissipation processes inside the battery as well as the temperature response. Common thermal models based on the physical mechanisms include the electro-thermal model, electrochemicalthermal model, and TR model. In additional, these models based on the dimensions can divide into the lumped model, 1D model, 2D model, and 3D model[35]. The
 following section introduces the heat generation mechanisms and heat transfer models
 of LiBs, as it is shown in Fig.1.

#### 4 2.1 Heat generation mechanism

The performance of LiBs is greatly influenced by temperature. Therefore, it is 5 crucial to grasp the internal heat generation mechanism of LiBs[36,37]. Under normal 6 working conditions, the internal heat generation of LiBs is usually inevitable due to 7 electrochemical reactions during charge and discharge cycles, and also affected by 8 9 environmental conditions; in the case of thermal runaway, the heat generation of LiBs comes from the undesired side reactions at high temperatures. The categories of heat 10 generation and dissipation within the battery system are depicted in Fig. 2. The open-11 circuit voltage (OCV) variation with temperature leads to the entropy change, 12 producing the reversible heat [38]; the irreversible heat mainly includes the 13 polarization heat, the heat of mixing, and the phase transition enthalpy. The 14 polarization heat is the heat loss caused by the ohmic polarization, the activation 15 polarization, and the concentration polarization. The heat of mixing comes from the 16 formation and relaxation of concentration gradient during the operation of LiBs. The 17 heat of mixing in the porous electrode can divide into four modes: 1) concentration 18 gradient distribution in spherical particles; 2) concentration gradient distribution in the 19 bulk electrolyte; 3) concentration gradient distribution in the electrolyte within the 20 pores of the electrode; 4) concentration gradient distribution in the bulk electrode. 21 Finally, the phase transition enthalpy that lithium ions insert and de-insert the 22 electrodes lead to entropic changes inside the electrodes is also one of the components 23 of irreversible heat [34,39]. 24



1

2

8

FIGURE 2 Summary of battery's heat generation and dissipation[40]

In the 1980s, Bernardi proposed a calculation model of the heat generation of LiBs assuming that the internal heating of the LiBs is uniform. In this model, the heat generation mainly divided into chemical reaction heat and ohmic heat generated by the internal resistance [41]. Additionally, they through experiment data simplified the LiBs heat generation equation as follows[42]:

$$q = I(U - V) - I \left(T \frac{dU}{dT}\right)$$
(1)

9 q, I, U and V respectively represent the heat generation, the electric current, the 10 open-circuit voltage and the cell voltage of the LiB. The establishment of this model 11 laid a foundation for future studies of the heat generation rate of LiBs. Most 12 computational models used for their following researches are the prototypes or 13 evolutionary forms of this model.

It is an urgent need for battery thermal management to refine the heat generation mechanisms of LiBs at both cell, module, aging and TR levels, especially the nonuniformity of heat generation, and to clarify the heat generation law of LiBs in their full-lifespan. Doh et al. pointed out that the internal heat generation of LiB includes three parts, namely the ohmic polarization heat, the reversible heat, and the irreversible heat. He noted that the heat generation of LiBs mostly depends on the applied current, the internal resistance, the temperature, and the entropy of the battery and indicated a method to measure the entropy. The adiabatic slow cooling strategy was used to
measure the variation of open-circuit cell potential with temperature, which could
evaluate efficiently the entropy of LiBs[43].

Wang et al. studied the ohmic resistance, the polarization resistance, 4 electrochemical impedance spectroscopy (EIS) and the heat generation of Li<sub>4</sub>Ti<sub>5</sub>O<sub>12</sub> 5 within two different life cycles (One was cycled 2000 times at 55°C and swollen, the 6 7 other is a fresh one). The results showed that the ohmic resistance of the swollen cell was higher than that of the new cell, but the polarization resistance was more minor. 8 The heat generated by the discharging process was greater than that caused by the 9 charging process. In the process of single charge and discharge, the heating rate of the 10 11 expansion battery was higher than the new battery [44]. The swollen cell released more reversible heat during the charging process, especially at a lower stage of charge. It is 12 consistent with the studies of Giel et al. They pointed out that battery aging impacts on 13 the calorific value and available capacity of LiBs under working conditions[45]. 14

According to the internal resistance joule heating model, Yoo et al. derived a 15 simple formula to calculate the irreversible heat of LiBs, which avoided the calculation 16 of complex electrochemical reactions. EIS measurement performed at small intervals 17 in the 0-10% and 90-100% State of charge (SOC) regions to capture the rapid changes 18 of the internal resistance of the LiBs. The analysis showed that the reason for this rapid 19 change is the intercalation of lithium ions into the electrodes. Furthermore, the heat 20 generation of a full-scale battery pack that composed of 280 cylindrical cells was 21 further analyzed. The analysis indicated that the temperatures of the tray predicted by 22 the simulation are consistent with the experimental results. Thus, the feasibility of the 23 thermal model for thermal analyses and the design of full-scale battery systems was 24 confirmed[37]. 25

Under battery abuse conditions, the heat generated from the side reactions is mainly produced after the TR. According to the order of reaction temperature rise, the side reactions of LiBs can classify as four categories: the decomposition reaction of solid electrolyte interface (SEI) film, the reaction of lithium with the binder in the
anode material, the reaction of the cathode material with the electrolyte, and the
decomposition reaction of the electrolyte[46,47]. Under routine runtime conditions,
heat generated from the side reactions is so tiny that it can neglect.

#### 5 2.2 Heat generation and heat transfer model of battery

Since the characteristics of LiBs are particularly sensitive to temperature, 6 revealing the thermodynamics of LiBs is crucial. The thermal models are often applied 7 to study the distribution of temperature fields in LiBs. The electrochemical model 8 9 quantitatively describes the micro-physical and chemical processes in the LiBs from the principle, which not only can accurately simulate the external characteristics of the 10 11 battery, but also can simulate the internal behaviors such as lithium-ion migration and diffusion in the electrode and electrolyte. The electrochemical model was first 12 proposed by Doyle, Fuller, and Newman. Based on the theory of porous electrode and 13 14 concentrated solution, they established the P2D porous electrochemical model[48–50]. In this model, it considered that the Lithium-ion transport between the anode and the 15 cathode is one-dimensional during the discharging and charging processes. They are 16 assuming that the active solid material is also composed of spherical particles. 17 Meanwhile, a series of partial differential equations (PDEs) are used to describe the 18 ion transport mechanism in the cell, the mathematical equations governing the charge 19 and mass conservation in solid and electrolyte phases is discussed in details in [51] and 20 summarized in Table 1[34]. 21

wherein, the chemical dynamic reaction process inside the battery is described by
the Butler-Volmer current density equation, it governs the volumetric rate of the
chemical reaction and is can be followed as[52]:

25 
$$j^{li} = a_s j_0 \left[ \exp(\frac{\alpha_a}{RT} \eta) - \exp(-\frac{\alpha_c}{RT} \eta) \right]$$
(2)

26 Where the overpotential  $\eta$  is defined as:

$$\eta = \phi_s - \phi_e - U(C_{se}) \tag{3}$$

2 and the coefficient  $j_0$  is a function of the surface electrolyte concentration  $C_{se}$  and 3 obtained as:

4

9

10

$$j_0 = (\mathcal{C}_e)^{\alpha_a} \left( \mathcal{C}_{s,max}^{n,p} - \mathcal{C}_{se}^{n,p} \right)^{\alpha_a} (\mathcal{C}_{se}^{n,p})^{\alpha_c} \tag{4}$$

5 Finally, the cell potential, V, across the cell terminals is determined as follows [53]:

6 
$$V = \phi_s(x = L) - \phi_s(x = 0) - R_f \frac{l}{A}$$
 (5)

7 The heat generation in cells mainly includes three parts, which can be followed by [54]:

8 
$$q_o = \sigma^{eff} \left(\frac{\partial \phi_s}{\partial x}\right)^2 + k_{eff} \left(\frac{\partial \phi_e}{\partial x}\right)^2 + k_D^{eff} \frac{\partial \ln c_e}{\partial x} \frac{\partial \phi_e}{\partial x}$$
(6)

$$q_p = j \ (\phi_s - \phi_e - U_i - \frac{j}{a}R_{film}) \tag{7}$$

$$q_{rev} = -\frac{j_{int}T\Delta S}{nF} \tag{8}$$

11 Where  $q_o$  is the ohmic heat,  $q_p$  is the heat generated by the over potential and  $q_{rev}$ 12 is the reversible heat.  $k_D^{eff}$  is the diffusional conductivity, n is the charge number of 13 lithium ion and  $\Delta S$  is the entropy change.

At present, the classical P2D model is widely used in studies such as battery aging mechanism, state estimation, and fast charging strategy. For example, Rahman et al. proposed a novel particle swarm optimization method to identify the specify parameters of a LiB based on the electrochemical model [55].

18 Table 1

Conservation equations	Boundary conditions	Initial conditions
Mass transfer in the solid phase		
$\frac{\partial c_s^{n,p}(x,r,t)}{\partial t} = \frac{D_s}{r^2} \frac{\partial}{\partial r} \left[ r^2 \frac{\partial c_s^{n,p}(x,r,t)}{\partial r} \right]$	$\frac{\partial c_s^{n,p}}{\partial r}\Big _{r=0} 0,$	$C_{S}^{n,p}(x,r,t_{0}) = C_{S_{0}}^{n,p}(x,r)$
Mass transport in the electrolyte	$\left. \frac{\partial C_s^{n,p}}{\partial r} \right _{r=R_s^{n,p}} = \frac{-j^{li}}{D_s a_s F}$	$C_e(x,t_0) = C_{e0}(x)$

$\frac{\partial \varepsilon_e C_e(x,t)}{\partial t} = D_e^{eff} \frac{\partial^2 C_e(x,t)}{\partial x^2} +$	$\frac{\partial c_e^p}{\partial x}\Big _{x=0} = 0, \ \frac{\partial c_e^p}{\partial x}\Big _{x=L} = 0$	$\phi_e(x,t_0)=\phi_{e,0}(x)$
$\frac{1-t_0^+}{F} j^{li}$ Potential in solid electrodes $\sigma^{eff} \frac{\partial^2}{\partial x^2} \phi_s(x,t) = j^{li}$	$\frac{\partial}{\partial x}\phi_{s}(x,t)\Big _{x=0,L} = \frac{-I}{A\sigma^{eff}}$ $\frac{\partial}{\partial x}\phi_{s}(x,t)\Big _{x=\delta_{n},\delta_{n}+\delta_{sep}} = 0$	$\phi_e(x,t_0) = \phi_{e,0}(x)$
Potential in electrolyte $k^{eff} \frac{\partial^2}{\partial x^2} \phi_e(x, t) +$ $k_D^{eff} \frac{\partial^2}{\partial x^2} \ln C_e = -j^{li}$	$\frac{\partial}{\partial x}\phi_e(x,t)\Big _{x=0,L}=0$	

Although the P2D electrochemical model could accurately describe the 1 2 electrochemical behavior of LiBs, there is a lack of thermal behavior calculation and 3 calibration of thermal model parameters. The thermal behavior of LiBs can be simulated by thermal models, for which the key is the calculation of the heat generation 4 and the temperature field distribution. There are two main types of heat generation rate 5 calculation models, namely the distributed heat generation model and the lumped heat 6 7 generation model[56,57]. Based on distributed heat generation model and P2D electrochemical model, an electrochemical-thermal coupling model was proposed[58-8 60]. This coupling model not only uses the simulation results of the electrochemical 9 model to calculate the heat generation rate and battery temperature, but also uses the 10 11 calculated temperature to modify the thermal-related electrochemical parameters of the electrochemical model. The electrochemical-thermal coupling model mainly uses the 12 Arrhenius equation (Eq. 9) or measured data to modify the thermal-related 13 physicochemical parameters such as solid-phase diffusion coefficient, liquid phase 14 diffusion coefficient, effective conductivity of the electrolyte, and the rate constant of 15 electrochemical reactions[31,34,61]. 16

17 
$$\Phi = \phi_{\text{ref}} \exp\left[\frac{E_a}{R} \left(\frac{1}{T_{\text{ref}}} - \frac{1}{T}\right)\right]$$
(9)

1 where  $\phi$  and  $\phi_{ref}$  are the values of the parameters under current and reference 2 temperature respectively, R is the molar gas constant, T and T<sub>ref</sub> are the 3 thermodynamic temperature and reference temperature,  $E_a$  is the apparent activation 4 energy.

The electrochemical-thermal coupling model describes the ohmic heat, 5 polarization heat, and entropy of LiBs from the perspective of electrochemical reaction 6 7 heat generation and mainly simulates the internal temperature of the battery under normal working conditions. Most of the electrochemical-thermal coupling models are 8 based on the conservation of material and energy, assuming that the current distribution 9 in battery is uniform. Due to the fact that the difference of current gradient inside the 10 11 battery is neglected, these models are only suitable for small batteries to ensure the accuracy of the simulation results. 12

The electro-thermal coupling model is suitable for studying the thermal safety 13 problems caused by the current inconsistency of LiBs. It is often based on the 14 electrochemical-thermal coupling model to improve the accuracy of the model. Based 15 on the principles of electron conduction, mass transfer, energy conservation, and 16 electrochemistry, Ye et al. established an electro-thermal coupling model to predict the 17 18 diffusion rate of lithium ions and the chemical reaction rate of cathode materials, which showed excellent agreement between the modeling results and experimental 19 results[62]. Chen et al. established the electro-thermal coupling model of the battery 20 pack and realized the accurate prediction of the battery pack temperature[63]. 21

Multi-field coupling models were developed. Zhang et al. proposed a typical sandwich model for the mechanical-electric-thermal coupled simulation of LiBs. This model predicted the initial temperature of TR caused by mechanical extrusion. In addition, they also explored the evolution law of voltage and temperature caused by mechanical extrusion[64]. Yang et al. indicated an electrochemical-thermalmechanical coupling model to study the aging mechanism of LiBs. Based on the previous electrochemical-thermal model, this model introduces the influence of
 mechanical stress on TR. The stress-stain equation is as follows:

$$\varepsilon_r = \frac{1}{E} \left( \sigma_r - 2\nu\sigma_\theta \right) + \frac{1}{3}\Omega C_s \tag{10}$$

3

$$\varepsilon_{\theta} = \frac{1}{E} [(1 - v)\sigma_{\theta} - v\sigma_{r}] + \frac{1}{3}\Omega C_{s}$$
(11)

5 Where *E* is Young's modules, *v* is Poisson's ratio,  $\Omega$  is the partial molar volume 6 of the solute,  $\varepsilon_r$  and  $\varepsilon_{\theta}$  are the radial and tangential strains,  $\sigma_r$  is the radial stresses, 7  $\sigma_{\theta}$  is the tangential stresses. The model considered not only the side reactions of the 8 anode but also the loss of active materials of the cathode. The aging behavior of LiBs 9 under different discharge rates and ambient temperatures was studied using this model, 10 which showed that the main reason for battery aging under high discharge rates is the 11 loss of active materials [54].

12 Studies on the heat transfer of LiBs are mainly concerned with the temperature 13 distribution inside the battery and the heat dissipation between the battery and the 14 ambient environment. The heat transfer equation as shown in (Eq. 12)

15 
$$\rho C_P(\frac{\partial T}{\partial t} + v_e \cdot \nabla T) \approx \frac{\partial (p C_p T)}{\partial t} = \nabla \cdot \lambda \nabla T + q \qquad (12)$$

Therein,  $\rho$  expresses the average density of the battery.  $C_P$  expresses the 16 17 average heat capacity under constant pressure.  $v_e$  is the electrolyte velocity.  $\lambda$  is the 18 average thermal conductivity in all directions and q is the heat generation rate[30]. At present, most studies assume that the cell is solid, ignoring the heat transferred by the 19 20 convection between the electrolyte and the porous electrode. The external heat conduction of LiBs mainly takes into account that the heat transfers between the LiBs 21 22 and high thermal-conductive media, such as the heat transfer between the cold plate and the battery, and heat transfer between the cold plate and the cooling medium. Most 23 studies ignore the radiation effect of LiBs on the ambient environment, which is 24 reasonable since the actual battery pack is often surrounded by thermal insulation 25 materials[65,66]. To study LiBs heat transfer, it is difficult to use analytical methods to 26

solve the complex heat transfer process coupling multi-physical phenomena, such as 1 solid heat conduction differential equation, Navier-Stokes equation, phase change, and 2 soon. Therefore, the finite element method is often used to solve these problems. 3 However, this method consumes massive computational resources and is not suitable 4 for online applications[67,68]. Some scholars proposed the equivalent thermal 5 resistance network model to study the heat transfer of the LiBs[69-72]. In this method, 6 the similarity principle is used to simulate the physical processes with equivalent 7 8 circuits. The amount of calculation is significantly reduced, but the distribution of the battery current field is often ignored. Hence the accuracy of the result is limited[73]. 9

The thermal safety of LiBs limits EVs' applications. TR is often triggered by 10 11 various abuse, including thermal abuse, electrical abuse, and mechanical abuse[74]. Some extreme environmental conditions may also cause high internal battery 12 temperatures and trigger a series of side reactions. The accumulation of heat generated 13 by the side reactions will make the temperature rise sharply, which then will further 14 aggravate the process of side reactions and eventually lead to TR safety problems. The 15 16 thermal abuse model plays a part in understanding the mechanisms of TR, improving vehicle safety, and TR warning. LiB thermal abuse models are generally based on the 17 traditional electrochemical-thermal coupling model, along with possible TR side 18 reactions and the associated heat generation within the battery. These TR thermal 19 models can simulate and predict the evolution process of TR under the condition of 20 thermal abuse. According to model proposed by Hatchard [75], the total heat generation 21 rate of electrochemical side reactions during the thermal runaway process is 22 approximately the Arrhenius equation as follows: 23

24

$$q_{de} = V\Delta H c^n A exp(-E_a/RT) \tag{13}$$

25 Where V is the volume of component substances of a LiB,  $\Delta H$  is the heat 26 generated by electrochemical side reactions, C is the reactant's concentration 27 involved in electrochemical side reactions, n is the reaction order,  $E_a$  is the 28 activation energy, R is the universal gas constant, and T is the reaction temperature. The common heat generation equations in the process of TR are summarized in Table
 2.

3 Additionally, Qu and Rao et al. conducted in-depth researches in the field of thermal safety, and recently summarized the inducements of TR over a wide 4 temperature range and also proposed relevant solutions[76]. Feng et al. introduced a 5 TR propagation model, which is consistent with the experimental results and could 6 7 provide corresponding solutions for suppressing TR[77]. To investigate the TR 8 propagation behaviors in the battery pack, Jiang created a lumped thermal resistance network, which is based on the heat transfer characteristics of LiB packs. In their study, 9 10 TR propagation features were discussed with different TR trigger locations, and the 11 TR prevention effect of PCM was evaluated. The proposed prediction method shows a good ability to resolve the TR propagation problem [78]. 12

#### 13 **Table 2**

Thermal runaway reaction equation[79].

Reaction stage	Reaction	Reaction equation
	temperature	
Solid electrolyte interface (SEI)	70°C	$Q_{sei}(T_b, C_{sei}) = H_{sei}W_{sei}a_{sei}exp\left(\frac{-E_{sei}}{RT}\right)C_{sei}$
decomposition		$\frac{\mathrm{d}C_{\mathrm{sei}}}{\mathrm{d}t} = -a_{\mathrm{sei}}exp\left(\frac{-E_{sei}}{RT}\right)C_{sei}$
Reaction of negative-electrolyte	120°C	$Q_{ne}(T_b, C_{ne}) = H_{pe}W_{ne}a_{ne}exp\left(\frac{-E_{ne}}{RT}\right)exp\left(\frac{-z}{z_0}\right)C_{ne}$
		$\frac{dC_{ne}}{dt} - a_{ne} \exp(\frac{-E_{ne}}{RT}) \exp(\frac{-z}{z_0}) C_{ne}$
		$\frac{dz}{dt} = a_{ne} \exp(\frac{-E_{ne}}{RT}) \exp(\frac{-z}{z_0}) C_{ne}$
Reaction of positive-electrolyte	200°C	$Q_{pe}(T_b, C_{pe}) = H_{pe}W_{pe}a_{pe}\exp\left(\frac{-E_{pe}}{RT}\right)\alpha (1-\alpha)$
		$\frac{dc_{pe}}{dt} = a_{pe} \exp\left(\frac{-E_{pe}}{RT}\right) \alpha (1-\alpha)$
Reaction of electrolyte	230°C	$Q_e(T_b, C_e) = H_e W_e a_e exp\left(\frac{-E_a}{RT}\right) C_e$
decomposition		$\frac{dC_e}{dt} = -a_c exp\left(\frac{-E_e}{RT}\right)C_e$

#### **3.** Application of BTMS in EVs

BTMS has many components as shown in Fig.3, including sensors, controllers, 2 actuators and the primary functions have rapid cooling, efficient preheating, TR 3 treatment and communication etc. The goal of BTMS includes temperature reduction, 4 uniform temperature distribution, reduction of energy consumption, economic costs, 5 and improved safety. The function of BTMS plays a vital role because it can protect 6 the battery by always maintaining a safe working range to reduce side effects such as 7 battery power and capacity decline. The detailed data of the battery, such as terminal 8 voltage and battery current, are captured by the cell monitoring function. The data 9 10 collected in real-time can predict the state of the battery in the next stage[40]. In the following part of this section, which mainly introduces the application of thermal 11 management in EVs, including cooling methods, preheating at low temperature, and 12 TR treatment. Additionally, the comparison of several different thermal management 13 14 technologies is summarized in Table 3.



FIGURE 3 Multi-objective optimization scheme for BTMS[77,80,81]

#### **3 3.1 Cooling methods**

1

2

BTMS can divide into four categories as depicting in Fig. 4: air cooling, liquid
cooling, PCM thermal management, and thermoelectric element cooling. PCM thermal
management includes solid-liquid phase change, gas-liquid phase change, and heat
pipe (HP) cooling. Air cooling mainly focuses on natural cooling and forced air cooling.
Cooling of thermoelectric elements focuses on Peltier effect cooling.

#### 9 3.1.1 Air cooling system

Air cooling has several advantages, such as simplified accessories, lightweight and convenient maintenance. It is because of these advantages in the field of EVs that air cooling systems have been widely used. [82], for example, Mazda Demio EV,

Toyota Prius, Honda Insight, Nissan leaf, and BYD F3DM[40]. Nissan leaf pure EV 1 adopts a passive battery pack thermal management system. With the optimized 2 electrode design, the internal impedance of the LiB and the heat generation rate 3 reduced simultaneously. At the same time, the thin layer (the thickness of the cell is 4 7.1mm) structure makes the heat inside the battery less to accumulate, which reduces 5 6 the demand for the complex thermal management system[83]. The cooling capacity of the battery air cooling system can be significantly improved by integrating the battery 7 8 air conditioning system with the vehicle air condition system. Yet, it needs to coordinate the different environment requirements of the battery and the passenger 9 10 compartment. Tao et al. proposed a forced convection air cooling intelligent control thermal management system integrated with the vehicle air conditioning system. The 11 system established a lumped parameter cylindrical battery thermal model and used 12 Kalman observer to estimate the transient surface and internal temperature changes. 13 The inlet cooling air temperature of the battery was optimized using the optimal control 14 15 theory, and the model predictive controller was developed to adjust the refrigeration 16 compressor in order to track the ideal cooling air temperature[84].



#### FIGURE 4 Cooling methods for LiBs[12,26,67,85–90]

17 18

#### 1 3.1.2 Liquid cooling system

In the application of high-performance EVs, air cooling cannot meet the demand 2 for high-power battery heating and cooling. Because the specific heat capacity and 3 thermal conductivity of liquid are greater than that of air, the thermal management 4 system with liquid as the medium has been widely adopted. In general, liquid-based 5 BTMS has high thermal conductivity and can be classified into direct-contact and 6 indirect-contact cooling according to whether the battery contacts the cooling medium 7 directly[40]. The first method is mainly referring to immersing the battery in the 8 cooling oil. This cooling method is widely used in electronic components. The second 9 method is achieved via the contact of the battery with a cold plate filled with the 10 cooling liquid for heat dissipation. In this way, the battery is cooled by the cooling 11 cycle composed of the pump, valve, and pipeline. Although direct contact mode has 12 higher thermal performance, direct contact between the battery and the cooling oil is 13 not practical in vehicle battery modules. In addition, the viscosity of indirect contact 14 coolant such as glycol/water is lower than that of the cooling oil. A higher flow rate 15 can be obtained by the fixed pump. Therefore, the indirect contact method has been 16 17 widely adopted and become the major cooling method for commercial EVs. Tesla EVs adopt a liquid cooling system, which uses rows of S-shaped cooling isolation plates to 18 increase the contact area between the cells and the cold plate[91]. GM volt EV also 19 20 adopts a liquid cooling method, which is a typical liquid cooling system configuration 21 of the soft pack or square battery. In this system, between each pair of batteries arranged a cold plate, microchannel placed on the cold plate, and manifolds set on both 22 sides to provide coolant for the cold plate. The pressure loss in the cold plate is much 23 24 greater than the manifold pressure[92].

The research of liquid cooling systems mainly includes three aspects: cooling system configuration, heat flow control parameters, and new types of coolant. E et al. proposed a BTMS with a rectangular channel cold plate[93]. The modeling results showed that the best combination of parameters was: 45 mm for the channel width, 5

mm for the channel height, 4 for the channel number, and 0.07 m/s for the cooling 1 2 liquid flow rate. On this basis, the average temperature and temperature difference of the cooling plate is the smallest. Mohammed et al. presented a dual-purpose cooling 3 plate for prismatic batteries, which can prolong the service life and safety of batteries 4 in vehicle applications[87]. The operating temperature of batteries can keep within a 5 6 suitable temperature range through the cooling plate. On the one hand, the measure could increase the battery life. On the other hand, it could effectively control the heat 7 generated during the TR. The system adopted a coolant with 60% ethylene glycol in 8 water. Under the condition of 20°Cand 0.2L/min, the surface temperature of the battery 9 10 could keep below 25°C. And it can control the temperature of the battery in the TR period in 30 seconds to about 75°C when the coolant is 20°C and the coolant flow rate 11 12 is 30L/min. In addition, new cooling fluids such as nanofluids and metallic fluids have gradually attracted great interest from relevant scholars due to their advantages, such 13 14 as high thermal conductivities and low power consumption. Kiani et al. selected three nanoparticles-alumina (Al2O3), copper oxide (CuO), and silver oxide (AgO) for the 15 preparation of properly stable nanofluids to analyze the BTMS for pouch LiBs 16 modules[94]. Test results demonstrated that the suspended particles in the nanofluid 17 causes increasing in its thermal conductivity, hence, the existence of suspended 18 particles in cooling fluid enhances the heat transfer efficiency. Although liquid cooling 19 20 can control the temperature of LiBs effectively, its system design is more complex to protect the pack from liquid leakage and increases power consumption[95]. 21

#### 22 **3.1.3 PCM cooling system**

PCM brings convenience to the implementation of BTMS. It can absorb a lot of heat in the process of phase transition while keeping the phase change temperature unchanged for a long time[96,97]. This system has the advantages of passively buffering high temperatures and extending battery life[98]. Phase transition thermal management system can divide into solid-liquid phase change type and gas-liquid phase change type according to the PCM. The gas-liquid phase change includes direct

#### 1 cooling and HP cooling.

The solid-liquid PCM BTMS was first proposed by Al hallaj in 2000[99]. Many 2 3 researchers have indicated that solid-liquid phase transition thermal management is an efficient and feasible low-cost thermal management technology[100-103]. When the 4 battery temperature is below the phase transition point, the PCM can release heat to 5 maintain the battery temperature, thus achieving thermal management. This method 6 does not need additional circulating refrigeration equipment. However, the low thermal 7 conductivity, the drift of phase change point, and the deformation caused by thermal 8 stress after multiple thermal cycles of PCMs limit the application of the solid-liquid 9 phase change method. To improve characteristic of thermal management, nanoparticles, 10 11 fins and porous metal foam are adopted beside the PCM. With the two heat generation rates (4.6W, 9.2W), though comparing with adding nanoparticles into the PCM, using 12 different numbers of fins, and employing the metal foam on the thermal management 13 system performance are researched. The experiment found that the porous-PCM 14 compound present more excellent cooling performance than the nanoparticle-PCM and 15 fin-PCM ones, the average temperature of the cell decreased by 4K and 6K[104]. The 16 17 other research applying PCM (n-eicosane) combined with nanoparticles volume fractions of 5% Al<sub>2</sub>O<sub>3</sub> and three horizontal radial copper fins can obviously improve 18 19 the melting time up to 28.3%, it greatly improves the issue of low thermal conductivity of the PCM[105]. Wu and coworkers showed an innovative and facile BTMS based on 20 21 flexible composite PCM, therein, the composite PCM involving paraffin, olefin block copolymer (OBC) and swollen graphite. The OBC and swollen graphite was applied 22 as the supporting component and thermal conducting component, respectively. The 23 mentioned BTMS presented an excellent thermal control performance, When the 24 battery is discharged from 100% to 0%, it shows higher cooling performance than that 25 without PCM, particularly at high discharge rate (2.5C)[106]. The difference between 26 27 thermal management using composite PCMs and traditional thermal management methods, the thermal and mechanical properties of composite materials, and the 28 application of composite PCMs in the thermal management of batteries are generally 29

studied[107–110]. A novel thermal management strategy is proposed to enhance the 1 mechanical molding property and prevent paraffin leakage to some extent[88]. The 2 BTMS based on the composite materials of swollen graphite and paraffin and 3 polyethylene configured with low fins has the high surface heat transfer capability. 4 Additionally, the composite PCM indicated much better mechanical properties and 5 cooling effect in comparison to swollen graphite and paraffin composition and air 6 cooling. At the discharge rate of 3.5C, the as-constructed battery pack based on 7 composite PCM effectively keep battery temperature within 50°C and temperature 8 difference of 5°C for LiBs. However, PCM can only absorb heat passively. Under 9 10 extreme conditions, such as high ambient temperature and high heat flux, the depletion rate of effective phase change enthalpy is speedy. Once it is exhausted, the thermal 11 12 management system may fail. The hybrid cooling methods based on PCM combined with other cooing strategy, such as liquid cooling and HP system, can effectively 13 alleviate the above issues. Research constructed the water-based N-eicosane particles 14 (130nm) emulsion flow in two multi-channel heat sinks with eight parallel and 15 divergent rectangular mini-channels. Therein, the channel was installed at an angle of 16 2.06°. It is found that compared with pure water, nano-PCM emulsion can obviously 17 increase the heat transfer performance of heat sink. With  $Re_{bf} = 295$  (i.e., Reynolds 18 number of base liquid) and pump power is 3.21 W/cm<sup>2</sup>, the heat transfer enhancement 19 20 is up to 13.8% in the divergent mini-channel heat sink[111]. Additionally, the PCM (Neicosan) was added in the mini-channel cold plate as hybrid mini-channel cold plate 21 that has also improved the heat transfer rate, under the heat generation rate is 400 22 kW/m<sup>3</sup> and the water inlet is 0.01 m/s with 298.15 K temperature, by comparing the 23 mini-channel cold plate whether contains PCM to analyzed the cooling performance. 24 The results shown that the temporal average of the maximum battery temperature in 25 hybrid cooling system are 0.06, 1, and 10.35K lower in comparison to mini-channel 26 cold plate cooling in Q = 100, 200, and 400 kW/m<sup>3</sup>, respectively[112]. In terms of 27 hybrid thermal management optimized method, [79] applied the Kriging-High 28 dimensional model representation (HDMR) method to establish a surrogate model of 29

the hybrid BTMS. Therein, the hybrid BTMS based on PCM, liquid cooling and HP is 1 researched. Moreover, the main sensitive factors and temperature difference of battery 2 are determined by global sensitivity analysis. And the multi-objective particle swarm 3 optimization is adopted as the optimization strategy for thermal management system. 4 Under the 5C discharging-1C charging process and TR conditions, the optimized 5 hybrid BTMS has the best heat dissipation characteristic with a highest cell 6 temperature of 30°C and the lowest temperature difference (2.31°C). The optimized 7 results shown that the surrogate model based on Adaptive-Kriging-HDMR strategy can 8 guarantee high calculation accuracy and decline the hybrid BTMS of calculation cost. 9 10 In conclusion, through PCM combining other cooling strategy can effectively alleviate the low latent heat issues of PCM. 11

The HP is a kind of high-performance thermal transfer element based on gas-12 liquid phase change. It uses the evaporation and condensation of refrigerant to transfer 13 heat. It has high directional thermal conductivity, and its thermal conductivity is higher 14 than any known metals. As a result, it has been widely used in many industrial 15 16 fields[113]. Gan et al. proposed an equivalent thermal resistance model for the BTMS of HP[114]. Liang proposed a battery module thermal management system using HP 17 and discussed its thermal performance and electrochemical performance through 18 experiments and models. The research results indicated that HP has a good application 19 prospect for battery cooling[115]. The HP also can couple with other kinds of cooling 20 systems such as PCM. As reported in literature[116], the as-constructed HP-assisted 21 PCM based battery thermal management is feasible and effective with a relatively 22 longer operation time and more suitable temperature. Compared with PCM based 23 24 modules, the HP-assisted PCM battery module operates longer when it reaches the set temperature of 50°C.Additionally, a sandwich structure consisting of a battery, PCM, 25 and HP was proposed by Jiang. The lumped thermal model considered the PCMs 26 melting and the thermal response of the HPs, and revealed the underlying coupling 27 mechanism of battery temperature and phase change process at different environment 28

temperatures, different heat transfer coefficients at condensation section, and different
 thickness ratios of PCM and battery[117].

3 The refrigerant direct cooling system is a novel BTMS, which also uses gasliquid phase change to transfer heat[118-120]. This system can combine the air 4 conditioning system with the battery cooling system. It dramatically reduces the 5 complexity of the secondary liquid heat exchange cooling system with multiple 6 7 branches[26]. In the direct cooling thermal management system, the evaporator cold plate directly installs in the battery module. The refrigeration cycle is realized by 8 connecting the refrigerant circuit with the existing air conditioning system. The 9 evaporator cold plate and the air conditioner evaporator are in two different positions 10 11 in parallel refrigeration cycles[121]. As shown in Fig. 5, Shen et al. propose a new synergistic control of a direct cooling system with an on-board air conditioning system 12 to enhance the performance of battery thermal management. In particular, the system's 13 thermal response, energy efficiency, and irreversibility are analyzed based on the 14 vehicle system framework for high ambient temperature and high speed dynamic 15 16 conditions. A mathematical simulation model of the refrigerant-based BTMS was developed and analyzed its performance under different pressure conditions based on 17 AMESim. Under high temperature and high speed conditions, this direct cooling 18 system can effectively reduce the temperature rise of the cells ( $\approx 25^{\circ}$ C) due to the large 19 amount of latent heat that accompanies the phase change process. The temperature 20 uniformity between the cells can also be well ensured ( $\approx 3^{\circ}$ C)[122]. However, since 21 one refrigerant loop is connected to two evaporators simultaneously, the different heat 22 load requirements of the passenger compartment and the battery system may lead to 23 24 contradictions in thermal control strategies. If there is a conflict between these two requirements, the thermal comforts of passengers may be affected because for safety 25 reasons, the priority of using refrigerant is given to the BTMS[123]. Therefore, a multi-26 objective cooperative thermal control strategy is fundamental to balance the thermal 27 demands of batteries and passengers. In addition, since the compressor must 28

- 1 continuously operate regardless of the cabin air conditioning, the system may have a
- 2 high power consumption.



3

4

FIGURE 5 Diagram of the coupling thermal management system[122]

#### 5 3.1.4. Thermoelectric element cooling system

6 The cooling system of the thermoelectric elements have been widely studied for 7 its clean energy and environmentally friendly management[124]. Thermoelectric generators (TEG) can convert heat into electrical energy, and the excess heat can 8 9 convert to power other devices. The thermoelectric cooler (TEC) can convert electrical 10 energy into heat energy to provide heating and cooling for various electrical equipment. 11 In recent years, automobiles have already used the TEC for auxiliary functions. For example, the temperature control in the cabin of EVs and the heating and cooling seats 12 in a luxury car is also suitable for BTMS[125]. TEC consists of P-type and N-type 13 semiconductors matrix. This component uses current to pass through a circuit 14

composed of the TEC matrix to generate heat between the two semiconductors by the 1 Peltier effect [126]. Liu et al. indicated a novel BTMS that combing TEC with liquid 2 cooling and established a thermal model of a battery, and a TEC and calibrated the 3 4 model with experimental data[127]. Troler et al. researched the effect of artificially induced temperature gradients on cell performance by using TEC to maintain the 5 temperature gradient of the LiBs under isothermal and non-isothermal conditions[128]. 6 The thermoelectric thermal management system can accurately control the temperature. 7 However, it consumes a large amount of energy due to its low efficiency. 8

#### 9 Table 3

reference	Cooling	subject	Merit	Demerit
	methods			
Chen and	Air-cooling	Optimizing cell	The optimal system	The experimental
Song.[24]		spacing of pack in	performs good for	conditions are relatively
in 2019		the parallel air-	various inlet airflow	single, and many
		cooled	rates, high stability.	variable factors need to
				be determined.
Chen et al.	Air-cooling	The influence of	The experimental is	There are few
[25] in 2019		inlet region and	easy to implement and	comparative
		outlet region for	the model has high	experiments, more
		cooling	accuracy.	schemes should be
		performance		adopted for comparison.
Shang et al.	Liquid cooling	Thermal	Low computational	The influence of many
[19] in 2019		management	cost, BTMS has the	physical properties of
		system based on	high reliability.	coolant on the system
		liquid cooling		needs to be further
				studied.
Kiani et al.	Nanofluids	Application of	The comparative	The effect of
[98] in 2020	liquid-cooling	nanofluids in liquid	experiment is	nanoparticle diameter on
		cooling	representative and	the experimental results
			reliable	was not considered in
				this experiment.
Heyhat et al.	PCM cooling	Composite PCM	High model accuracy,	The effect of distribution
[108] in		use in battery	the experimental and	of nanoparticles in
2020		thermal	simulation results has	composite PCM on
		management	the excellent fitting	cooling performance
				needs to be further
				studied

The battery thermal management technology comparison

Zhang et al.	Hybrid cooling	Application of	High calculation	The experiment only
[80] in 2021		surrogate model in	accuracy, low	studied the cooling effect
		hybrid BTMS	calculation cost	of a single battery
Liao et	Thermoelectric			
al.[129] in	element	BTMS used TECs	Strong adaptability,	Complex structure, lack
2021	cooling		wide	of experimental
			operating temperature	verification
			range.	
Lyu et	Thermoelectric	TECs are used for	Rapid and accurate	There are many
al.[130] in	element	thermal control in	cooling strategy	interference factors in
2019	cooling	LiBs		the experimental content,
				and the experimental has
				limitations.

#### 1 3.2 Preheating works at low temperature

The performance of the LiBs will significantly reduce at low temperatures. Hence it is necessary to preheat the LiBs in EVs at low temperatures. Generally, preheating methods included external heating and internal heating, as presented in Fig. 6. The external heating system is usually the most common and simplest method[131,132] but requires the radiation heater and is complex in structure. Internal heating system can reach the predetermined temperature quickly, but the safety risk is high[133].



#### 2 3.2.1 External heating

For all types of batteries, air preheating is a suitable method and has been widely used in EVs [135–137]. However, the air circulation requires the configuration of the fan and flow channels, which will increase the complexity and reliability issues of the BTMS. The noise produced by the fans could cause the poor comfort of the driver and passengers to a certain extent. Moreover, because of the low thermal conductivity of air, whether it is passive or active air heating, this will extend the working time for preheating the battery.

10 The liquid preheating system has an ideal preheating effect due to its higher thermal conductivity and heat transfer rate, but the preheating system as a whole is 11 relatively complicated [138]. Most of the liquid preheating system includes heaters and 12 heat exchangers, among which the cell contacts the heat exchanger, and it preheats the 13 cell by flowing liquid through the heat exchanger. As shown in Fig. 7, under New 14 European Driving Cycle conditions, Wu et al. implemented a method for externally 15 16 preheating cells with liquid. The experimental battery pack could be heated from -30°C to 10°C within one hour and the maximum temperature difference between the cell 17 could be reduced to 1.6°C[139]. According to the contact type between the liquid 18 19 medium and the battery, liquid preheating BTMS could be classified as non-contacting 20 preheating and immersing preheating[140]. The non-contacting liquid preheating method has been applied to EVs. For example, Volt heats the liquid medium using 21 electric heaters, which flows into the whole battery pack[141,142]. In addition, Tesla 22 23 Motor also uses the liquid heating method to heat the battery pack. The flow circuit used in the system included four channels that allow the liquid medium to flow more 24 fully and shorten the preheating time[143, 144]. In general, the immersing preheating 25 has a higher thermal conductivity than the non-contacting preheating, so the battery 26 27 has a more uniform temperature distribution during the heating process. Nevertheless,

effectively preventing the leakage of the liquid medium is still a significant issue, and
the liquid medium is required to have good electrical insulation properties to reduce
the risk of the short circuits [134].



5

**FIGURE 7** The schematic of liquid-based BTMS[139]

Compare with air and liquid, PCMs can absorb and release heat from the external 6 7 environment during phase transitions. The melting process can apply to equipment cooling, and the solidification process can be used for low temperature heating[145]. 8 The PCM is generally isolated from the battery by the container. The advantage of 9 PCM-based BTMS is that it can keep the temperature uniformity of the battery. Hence 10 it is beneficial to prolong the lifespan of the batteries. Rao et al. conducted a 11 comparative simulation for PCM and air preheated batteries, and the results of the 12 13 present study indicated that as the thermal conductivity of the PCM increases, the temperature difference decreases. The heating time of air-heating was 6.4, 5.2, and 4.2 14 times of that of PCM-heating when the battery was heated from -30°C, -20°C, and -15 10°C to 10°C, with the initial temperature of air and PCM being both 50°C [146]. As 16 shown in Fig. 8, a battery preheating scheme for paraffin/graphite composite PCM was 17 proposed by Zhong et al. The cell in each module is wrapped with parallel resistive 18 19 wires, and then the remaining space within each module is filled with composite PCM. Five resistance wires can heat the battery from -25°C to 10°C in 273 seconds, 20 consuming 2948 J of energy[147]. 21



FIGURE 8 PCM-based preheating system for batteries[147]. (a) The battery module with resistance
wire twined around battery cells. (b) The battery pack consists of battery modules with CPCM
perfused in each module.

5 Besides, the preheating of the battery can be realized by electrothermal elements, 6 which include electrothermal films and electrothermal plates[148]. From the related 7 research, the electrothermal films can keep a better temperature distribution during the 8 heating process because of its performance, which is more suitable for the preheating 9 work of the battery pack and can effectively avoid the inconsistent temperature 10 distribution of the LiBs during the heating process[149]. Yet, whether this method 11 impacts LiB life remains to be studied[135].

#### 12 **3.2.2 Internal heating**

13 In general, the internal preheating methods of the battery include self-heating and 14 current excitation heating. Compared with the external heating method, the internal heating method has higher heating efficiency, and no requirements for cell shape[149]. 15 16 However, the heat generation mechanism is very complex, and improper use may cause 17 safety problems. As shown in Fig. 9, Wang et al. have developed a self-heating LiB 18 that could self-heat in low temperature ambient environment. The cell is heated by 19 inserting a nickel foil with two tabs into the LiB to generate heat. More notably, during the activation process, the surface temperature of the battery rises rapidly from -20°C 20 21 to  $0^{\circ}$ C in a matter of seconds[150,151]. Self-heating is an efficient strategy for preheating LiBs. But the self-heating effect of LiBs is limited by the physical properties
of the cell. In addition, most experimental data are only obtained from single battery.
For a battery pack with hundreds of individual cells, this cell modification could be
prohibitively expensive. Further, because the internal structure of the cell is changed,
it cannot use in current EVs due to safety concerns. The safety issues of the self-heating
method still need further study.



8 FIGURE 9 (a) The schematic of self-heating battery. (b) Cell voltage and temperature evolutions
9 during Vact=0.4V (inset) and subsequent 1C discharge at -20°C[151].

7

10 Current excitation can be direct current (DC) preheating, alternating current (AC) preheating, or pulse preheating. Heating the battery by discharging the battery at a 11 certain constant current is called DC preheating, which releases the energy stored 12 inside the battery. To avoid lithium plating and battery aging during preheating, it is 13 necessary to control the current amplitude and preheating time within a certain 14 limit[152,153]. By increasing the discharge rate and lowering the cut-off voltage of 15 DC preheating, i.e., prolonging the discharging time, the heating rate can be 16 increased[154]. 17

AC preheating method applies AC current of a specific frequency and amplitude to the cathode and anode of the cell, and the heat is generated from the internal impedance[155,156]. Compared with the DC preheating method, the AC preheating method can more quickly and more effectively heat the battery [157]. Zhang et al.

adopted a lumped energy conservation model to predict the battery temperature under 1 AC preheating. They preheated the battery with a sinusoidal AC at different 2 frequencies and current at low temperature. The results showed that the preheating time 3 decreased with the increase of the AC amplitude and the decrease of the frequency. 4 Better thermal insulation of the battery also helped to reduce the preheating time[158]. 5 6 It should be noted that the AC heating effect will reduce with the constant amplitude, when the frequency is too low. However, this study did not show how to optimize the 7 8 AC amplitude and frequency to improve the heating rate. Besides, the impact of the applied AC current on battery health is still unclear. In addition, the AC preheating 9 10 method requires an external power supply, which increases the cost or limits the application scenarios and may be the reason why it hasn't been used 11 12 commercially[135,147].

13 Pulse preheating method uses pulse current to generate heat through the internal impedance of the cell to preheat the batteries. Compared with air preheating, the pulse 14 15 preheating method can achieve a more uniform distribution of heat generation inside the battery pack, and a low temperature variance is beneficial to battery health [159]. 16 An effective preheating strategy is developed by using a thermo-electric coupling 17 model based on the EIS of the cell and the pulse frequency is varied between 1000Hz 18 and 3637.1Hz to compare their preheating effect. The optimal pulse configuration (the 19 2637.5Hz frequency) can heat the cell from -20°C to 5°C within 1000 seconds. The 20 battery's performance temperature is significantly improved after this temperature rise. 21 22 Further, experimental data show that the damage to the battery health, in terms of 23 capacity attenuation and internal impedance rise, is minor[160]. However, this method is only validated in the laboratory, and its effectiveness has not been verified in the 24 actual vehicle. 25

#### 26 **3.3 Thermal runaway treatment**

The mechanical, electrical, and thermal abuses (overcharge, over-temperature and metal penetration, etc.) of LiBs can cause the TR phenomenon, which will pose a safety hazard to the battery pack and passengers[158]. Battery TR, if not detected and interrupted in time, will cause serious safety accidents [26]. At the same time, due to the large specific heat of the battery system, it is difficult to quickly control the temperature to reach the predetermined temperature range when the TR occurred, so there is a lag problem in the thermal management system. For these issues, there are four categories of emergency battery thermal barrier (EBTB), including material optimization, the TR forewarning strategy, fire extinguishment, and heat removal.

8 Material optimization strategies include Electrolyte additives, separator safe and electrode design. Wang et al. proposed a new idea to enhance the battery safety by 9 passivating the cell first and then self-heating before use. They had proved that, adding 10 11 triallyl phosphate to the electrolyte, can improve the stability of LiB under high pressure and high temperature operating conditions[161,162]. Liu et al. demonstrated 12 a new type of separator that improves flame retardancy by adding triphenyl phosphate 13 to the conventional electrolytes. During LiBs TR, the protective polymer shell would 14 melt, releasing the flame retardant to suppress the burning of the electrolyte inside the 15 16 battery[163]. Chen et al. proposed a thermoresponsive polymer switch material, a mixture of graphene-coated sharp nickel nanoparticles. Once the TR phenomenon 17 occurs and the battery temperature rises above switching temperature (T<sub>s</sub>), the 18 conductivity of this material will drop sharply, thereby suppressing further exothermic 19 side reactions and the propagation of TR[30,163]. 20

Emergency cooling system needs to monitor the battery temperature and detect 21 the starting point of TR and then quickly remove the accumulated heat. The pre-22 warning of TR depends on the early detection of abnormalities in the measurements of 23 the battery's terminal voltage[164], mechanical deformation[165], internal 24 temperature[166], and internal impedance of the battery[167]. Additionally, the gas 25 component identification is another useful method for thermal runaway 26 27 prewarning[77]. Zhu et al. indicated that the voltage of LiBs would drop sharply before 28 TR occurs, and this characteristic can be used to warn the upcoming risk of TR[168]. After a TR warning, the battery pack must initiate the heat-dissipation procedure 29

immediately [169]. Once the TR or even an explosion hazard happens, firefighting and 1 heat removal must be carried out to cut the propagation routes and to control the 2 damage. Wang et al. studied using dry powder and water to extinguish the flame, and 3 showed that this measure can suppress the TR propagation of LiBs[170-172]. As 4 shown in Fig. 10, Liu et al. proposed a novel BTMS with combining fast water mist 5 6 and  $C_{6}F_{12}O$  cooling. They found out that this method could reduce significantly the peak temperature and the high-temperature duration of the cell after TR, as compared 7 8 to the method with C<sub>6</sub>F<sub>12</sub>O only and without suppression during the heatwave. After the TR is initiated, by combining the C<sub>6</sub>F<sub>12</sub>O and the water mist to spray onto the 9 10 battery surface, the cell can be effectively cooled [173]. Additionally, an optimized dual-functional battery module that involved PCM and aerogel is indicated[174]. The 11 12 PCM inset EG availably suppress the combustion flame and reclined the peak temperature of battery. Meanwhile the aerogel showed the outstanding performance in 13 14 delaying battery TR. The proposed BTMS combined the advantages of aerogel and PCM. The optimized module delays the initial time of TR by 173s and alleviates the 15 thermal propagation. Moreover, reducing the maximum temperature of battery module 16 to 29°C. Meanwhile, studies have shown that emergency spray flow has the limited 17 efficiency on reducing the temperature of TR battery, wherein the nozzle initial 18 condition is set to 0.2Mpa, 0.5L/min<sup>-1</sup> and the spray angel is about 60°. The fact is that 19 20 spray cooling can be adopted to the emergency cooling strategy to suppress TR[175]. In addition to mentioned methods, Gao et al. investigated in depth the emergency 21 cooling and overheating control of LiBs, and together with the direct cooling BTMS. 22 23 They proposed an open-loop emergency safety control method, namely, when the TR 24 state is detected, refrigerant is directly injected in the LiBs pack for emergency cooling and flame retardancy[176]. 25



1

FIGURE 10 (a) Schematic diagram of water mist penetrating plume. (b) Temperature response for
long surface of experiments extinguished and cooled by water mist with various working pressure[173].

In addition to the study of experimental mechanisms for triggering thermal 4 runaway, model-based TR predictions are essential for optimizing the safety design of 5 batteries. Jin et al. used COMSOL software to build a three-dimensional lithium-ion 6 battery model to investigate the combined effect of heating power and heating area on 7 triggering thermal runaway. The experiment adopted eight heating strategies, i.e., a 8 combination of two different heating powers and four incremental heating areas. The 9 results show that the area with a lower heating power density can trigger thermal 10 runaway faster with the same heating method. Based on the simulation results, 11 recommended strategies for thermal runaway prediction and low-temperature heating 12 solutions are proposed[177]. Ren et al. proposed a novel model for predicting thermal 13 runaway in lithium-ion batteries. Differential scanning calorimetry (DSC) tests on 14 15 individual cell components, and their mixtures were used to reveal and characterize the thermal runaway mechanism. Six main reactions (e.g., decomposition of the solid 16 electrolyte interfacial film) are used as the main heat generation sources. The kinetic 17 parameters of each exothermic reaction are determined using the Kissinger method and 18 a non-linear fitting approach. The model is in good experimental agreement with the 19 results of adiabatic TR tests and oven experiments for 24Ah Li-ion batteries. The 20 results show that the model can respond well to the thermal runaway mechanism of the 21 battery[178]. Sara Abada et al. developed a three-dimensional physical model of the 22

electrothermal behavior of lithium-ion batteries under thermal runaway conditions.
The combination of experimental and modeling analysis enables an excellent
understanding of the mechanisms of thermal runaway in Li-ion batteries and the
implications for battery aging. The experiment found that calendar aging leads to a
delay in the self-heating temperature of the battery so that thermal runaway can be
triggered at lower temperatures. Meanwhile, the experimental results are in good
agreement with the simulation model [179].

## 8 4. Connection and development of thermal management 9 system and CHAIN

10 Due to the large specific heat of the battery system, it is difficult to quickly control the temperature to reach the predetermined temperature range, namely, the thermal 11 management system exists lag problem. In addition, due to different triggering reasons 12 and the complexity of battery aging and environmental impact, pre-warning of TR is a 13 critical challenge[180]. Through the early prediction and pre-warning can alleviate 14 above issues. Combining BTMS with cloud computing to improve the computational 15 power and data storage capacity can enhance the response speed and TR early warning 16 ability. With the Internet of things (IoT), all battery relevant data can be measured and 17 transmitted to the cloud seamlessly, building up a digital twin for the battery system. 18 Diagnostic algorithms evaluate the data and enable optimization of the battery charging, 19 aging, and thermal management[181]. The CHAIN framework was proposed by Yang 20 et al. It is suggested that the critical physical and electrochemical parameters of battery 21 from production to use should be uploaded to the cloud server to optimize the whole 22 life management of the battery system. From the above-mentioned issues, a novel 23 24 solution for battery thermal management is proposed, which combines traditional BTMSs with the CHAIN architecture through the "end-edge-cloud" multi-layer 25 collaborative lay-out to achieve cloud-enhanced BTMS[182]. As shown in Fig.11, 26 CHAIN integrates the cyber-end and the vehicle-end, which can generate additional 27 functionality by adding new complex layers. Meanwhile, it may lead to potentially 28

complicating the design to guarantee performance on the cyber-end, while the single 1 vehicle-end carries out recording and uploading real-time data and executing 2 commands from the cyber-end via wireless communication[183]. In addition, Wu et al. 3 pointed out that a digital twin model of battery could be created, combining the 4 knowledge of battery degradation, modeling tools, and diagnosis with machine 5 6 learning. This network, through establishing the mapping relationship between the physical and digital embodiment of the battery, can achieve a more intelligent control 7 8 and a longer service life[184].





10

FIGURE 11 Multi-condition Control and Multi-Lifespan Recurrence[183]

Based on the points mentioned above, this paper presents a novel digital solution 11 for BTMS, as depicted in Fig. 12. The digital twin of battery is established to realize 12 the mapping of physical and digital entities. Some the basic data of battery are collected 13 14 at the vehicle end, mainly including current, temperature, acoustics, gas and battery or 15 module deformation. The data is transmitted to the cloud based digital twin through IoT technology, where data is collected and processed. The cloud based digital twin 16 17 feeds back the corresponding control parameters to the vehicle end, so as to realize accurate and efficient battery thermal management. Some basic functions and data are 18 19 executed and processed locally at the vehicle end, so as to reduce the computing load capacity of the cloud. The local model and algorithm processing are used to data driven of the vehicle end BTMS. Finally, based on the mirrored function of the digital twin framework, it can realize the functions of diagnosis and early warning, state estimation and data processing, it better serves the cloud based digital twin and the local computing at the vehicle end. The proposed digital solution described in this paper mainly includes the following parts: on-board sensing, local computing and cloud based digital twin.

#### 8 4.1 On-board sensing

9 The condition for the current intelligent BTMS is to accurately collect the relevant data of the system and infer the state of the system from these data. The data acquisition 10 depends on the sensing ability of the local terminal hardware. The information 11 collected by the battery system is as described above, such as current, voltage, 12 temperature, acoustics and deformation. Advanced IoT devices can provide stable and 13 14 fast computing, transmission and connection capabilities to effectively transmit data. Therefore, it is very important to develop advanced sensor technology to sense the 15 current state of the battery. As described in literature[185], mist computing can process 16 data at the extreme edge of the network, which can be realized separately on the current 17 Internet of things devices without communicating with the local computing or cloud, 18 so as to reduce the communication demand. 19

#### 20 4.2 Local-computing based on vehicle end

In specific cases, the ability of monitoring and processing information of the BTMS is very demanding on time. When the transmission communication and cloud computing capacity of the Internet of things are limited, edge computing is an effective way to reduce the efficiency of cloud load computing. Therefore, several thermal management functions be finished on vehicle-end, such as the efficient preheating, rapid cooling and TR treatment. In addition, there are also including the electrochemical model and thermal model computing. For improving the reliability of the thermal management system, the functions of each point during operation should be
 operated locally. Then the functions of data transmission, data synchronization and
 local computing are realized in the edge computing[186].

#### 4 4.3 The cloud based digital twin

with the amount of data generated, data aggregation, machine learning and 5 hybrid intelligent learning and control key algorithms be achieved in the cloud-end, 6 which the functions included data driven optimal, prognostics and digital health 7 metrics. Utilizing cloud computing and storage capabilities to conduct aggregation 8 9 analysis and feature extraction for the data in the cloud-end. Meanwhile, machine learning, artificial neural network and other algorithms are used to predict the battery 10 health state and heat generation, and conduct the fusion model for the battery 11 mechanism model and equivalent circuit model. The model algorithm is used to fuse 12 historical health data to provide pre-warning work of TR. 13

In addition, based on the CHAIN and digital twin framework, IoT and 5G 14 technology to realize the multi-scale and multi-stage prognostics and thermal 15 management. Finally, the mirrored system runs the advanced thermal model, diagnosis 16 and control algorithms to predict and evaluate the battery's SOX and internal 17 temperature. Meanwhile, it can also optimize the operating parameters, such as the 18 charging rate. These battery control parameters are then transmitted to the vehicle end. 19 With the greatly expanded computational capacity enabled by the cloud, advanced 20 battery management algorithms, such as data-driven and machine learning fleet 21 management tools[186] can be implemented. 22



1 2

FIGURE 12 Battery thermal management system leveraging from CHAIN [31,45,89,122,187,188]

#### **5. Conclusion and future task**

#### 4 5.1 Work summary

5 This paper provides a comprehensive review of the heat generation mechanism 6 and thermal models of LIBs, along with the existing battery thermal management strategies. A comprehensive discussion of the heat generation mechanism and heat 7 transfer models of batteries is presented, enabling an understanding of the approach 8 and importance of battery thermal management. Then, this is followed by introducing 9 10 current battery thermal management strategies, namely air cooling, liquid cooling, phase change materials, and electronic component cooling. Several comprehensive 11 analyses of the various thermal management strategies are investigated from model to 12 experiment. Finally, the paper proposes a CHAIN-based battery thermal management 13 14 system, which is explained in detail from three aspects: vehicle-side sensing, edge computing, and cloud-based digital twin. Finally, the paper aims to enable researchers 15 16 to understand the current thermal management strategy trend.

#### 17 5.2 Prospects

In terms of existing BTMS status, it has some simple functions such as open-loop
control, limited use of model/diagnosis/prediction capacity etc., additionally, no

intelligence or predictive control and difficulty of maintenance/repair. With the 1 continuous development of communication technology, the future BTMSs have the 2 smart battery pack with embedded sensors, new efficiency thermal management 3 approaches and telematics system 5G etc. Intelligent BTMS enabled by cloud-4 computing can fused big data, advanced diagnosis and optimization algorithms. Such 5 as digital twin and machine learning, optimal/predictive maintenance and repair, data-6 driven fleet management can be implemented in the smart BTMS. Furthermore, BTMS 7 will not only ensure the battery keeps more safe, efficient, and reliable conditions, but 8 communicate and collaborate in other fields, such as microgrid. The future task 9 10 followed as by:

- Developing advanced sensing technology, such as fiber-optic sensing, sensor
   based piezoelectric/pyroelectric poly and a thin-film transistor array and
   wireless sensor.
- Fusing the advanced data model algorithm by linking the historical data and
  real-time data to feedback, modifying the thermal model to provide a more
  accurate heat generation prediction.
- Predicting and pre-warning technology of battery heat generation and TR
   combined with CHAIN and digital twin to avoid TR events.

#### **19 Declaration of Competing Interest**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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